

Motivation

- Soot formation is a key process in flames and fires
 - Radiative heat transfer
 - Incomplete combustion
 - Environmental and health impacts
- Soot affects radiation, which affects temperature, which affects soot...
- Modeling soot formation is challenging
 - Complex chemistry among many species
 - Representation of the particle size distribution
 - Potentially several PSD coordinates
 - Large range of scales
- Combustion simulations
 - complex chemistry, turbulent flow, multicomponent mass transfer, radiative heat transfer, soot formation
 - Burden on users/developers for submodel expertise
 - Availability of libraries that offload submodels facilitates code development and progress
 - Chemkin, Cantera, etc.



<https://energy.sandia.gov/programs/nuclear-energy/nuclear-energy-safety-security/>

Soot Processes: Nucleation

Semi-empirical

Leung et al.

- K. M. Leung, R. P. Lindstedt, W. P. Jones, *A simplified reaction mechanism for soot formation in nonpremixed flames*, *Combustion and Flame* 87 (1991) 289–305
- $C_2H_2 \rightarrow 2C_{soot} + H_2$

Lindstedt

- R. P. Lindstedt, *Simplified soot nucleation and surface growth steps for non-premixed flames*, in: H. Bockhorn (Ed.), *Soot Formation in Combustion*, no. 59 in Springer Series in Chemical Physics, Springer-Verlag Berlin Heidelberg, 1994, pp. 417–441
- $C_2H_2 \rightarrow 2C_{soot} + H_2$
- $(C_6H_6 \rightarrow 6C_{soot} + 3H_2)$

Detailed

PAH Nucleation, Blanquart & Pitsch

- G. Blanquart, H. Pitsch, *A joint volume-surface-hydrogen multi-variate model for soot formation*, in: H. Bockhorn, A. D'Anna, A. F. Sarofim, H. Wang (Eds.), *Combustion Generated Fine Carbonaceous Particles*, KIT Scientific Publishing, 2009, pp. 437–463.
- 8 PAH species
- Free molecular PAH collisions to form PAH Dimers
 - $PAH_i + PAH_i \rightarrow \text{Dimer}$
 - Only self-collision assumed
 - Dimers not distinguished
- Dimer coagulation to form soot
 - $\text{Dimer} + \text{Dimer} \rightarrow C_{soot}$
 - Dimers assumed in steady state, quadratic eqn:

$$\underbrace{\sum_i^{n_P} \gamma_i \frac{1}{2} \beta_{P_i, P_i} n_{P_i}^2}_{\text{formation}} = \underbrace{\beta_{D, D} n_D^2}_{\text{soot nuc.}} + \underbrace{\sum_k^{n_s} \beta_{D, C_k} n_{C_k} n_D}_{\text{condensation}}$$

Soot Processes: Growth

Semi-empirical

Leung et al.

- $\text{C}_2\text{H}_2 + \text{C}_{n,\text{soot}} \rightarrow \text{C}_{n+2,\text{soot}} + \text{H}_2$
- $R_g \propto \sqrt{A}$

Lindstedt

- $\text{C}_2\text{H}_2 + \text{C}_{n,\text{soot}} \rightarrow \text{C}_{n+2,\text{soot}} + \text{H}_2$
- $R_g \propto A$

PAH Condensation

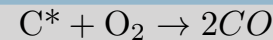
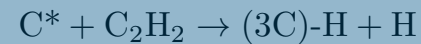
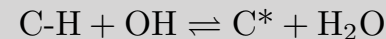
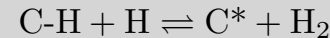
- Consistent with PAH nucleation
- $\text{D}_{\text{C}_x\text{H}_y} + \text{C}_{n,\text{soot}} \rightarrow \text{C}_{n+x,\text{soot}} + (y/2)\text{H}_2$
 - Dimer size computed as weighted average over PAH contributions to PAH nucleation

$$R_{\text{cnd},k} = \beta_{D,C_k} n_{C,k} n_D m_D \quad (=) \quad \text{kg/m}^3\text{s}$$

Detailed

HACA

- *ABF mechanism: Appel, Bockhorn, Frenklach, Combustion and Flame 121:122-136 (2000).*



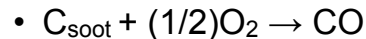
- C^* from QSSA over given reactions
- *Reverse reactions from Ken Revzan and Frenklach 02/15/02 code soot.f:*
combustion.berkeley.edu/soot/codes/routines.html
- *Rate is proportional to frac. available surf. sites α*

$$\alpha = \tanh \left(\frac{a}{\log_{10}(M_1/M_0)} + b \right)$$

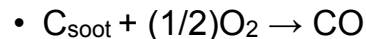
- $a(T)$, $b(T)$ from Balthasar and Frenklach, *Combust. Flame* 140:130-145 (2005)

Soot Processes: Oxidation

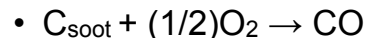
O₂ Global: Leung, Lindstedt, Jones 1991



O₂ Global: Lee, Thring, Beer 1962



O₂ Elementary: Nagle, Strickland-Constable 1962



- Graphite rods

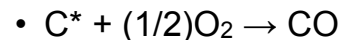
OH Elementary Neoh, Howard, Sarofim 1981



$$R = 1290 \times 0.13 P_{OH} T^{-0.5}$$

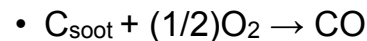
- R (=) kg/m²s, P_{OH} (=) atm, T (=) K

HACA O₂, OH: Appel, Bockhorn, Frenklach 2000



- Same rate as used in HACA for C* from QSSA

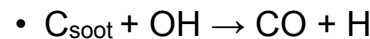
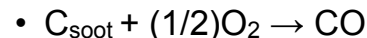
Optimized O₂, OH: Guo, Anderson, Sunderland 2016



- Optimized among 12 experiments.

- OH efficiency = 0.1

Optimized O₂, OH: Josephson et al. 2017



- Optimized among 13 experiments using Bayesian statistics.

- OH efficiency = 0.15

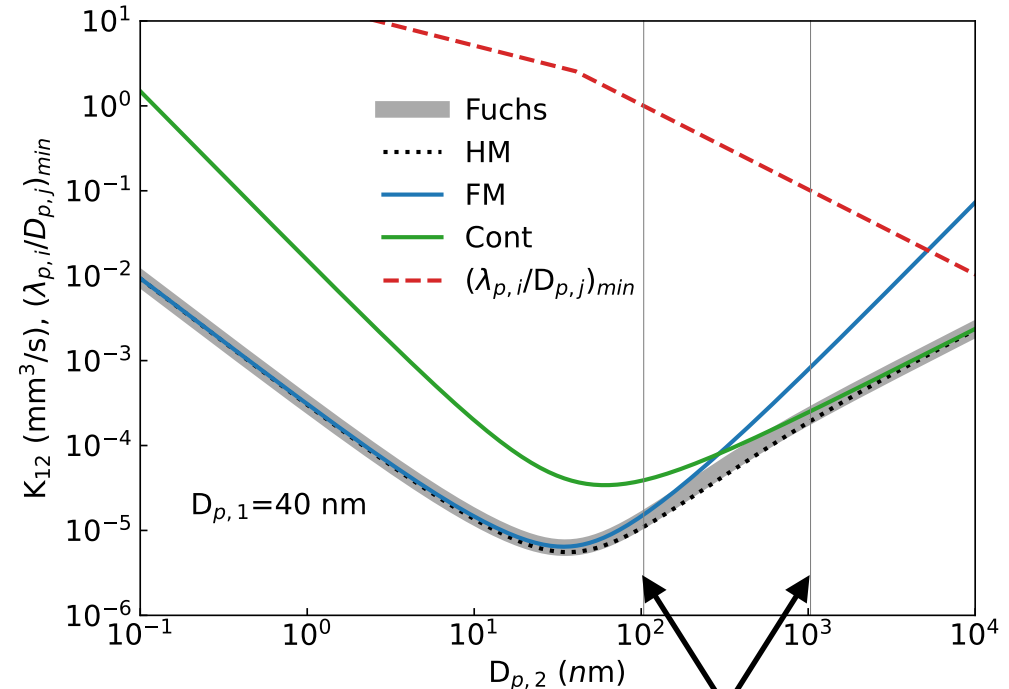
Soot Processes: Coagulation

- Assume spherical particles
- Two regimes
 - Free molecular: small particles
 - $D \ll \lambda_{mfp}$
 - Continuum: large particles
 - $D \gg \lambda_{mfp}$
- Note, λ_{mfp} is the mean free path of the particle, *not* the gas.

$$\dot{S}_{coag}(m) = \frac{1}{2} \int_0^m \beta_{\mu, m-\mu} n(\mu) n(m) d\mu - \int_0^\infty \beta_{\mu, m} n(\mu) n(m) d\mu$$

$$\beta_{m,\mu}^{FM} = c\epsilon_c \left(\frac{\pi k_b T}{2} \right) \left(\frac{6}{\pi \rho_s} \right)^{2/3} \left(\frac{1}{m} + \frac{1}{\mu} \right)^{1/2} (m^{1/3} + \mu^{1/3})^2$$

$$\beta_{m,\mu}^C = \frac{2k_b T}{3\mu_v} \left(\frac{C_m}{m^{1/3}} + \frac{C_\mu}{\mu^{1/3}} \right) (m^{1/3} + \mu^{1/3})$$



$$Kn = \left(\frac{\lambda_{p,i}}{D_{p,j}} \right)_{min} \quad 0.1 \leq Kn \leq 1$$

Brackets the transition region

Transition Region

- Harmonic mean $HM = \frac{C \cdot FM}{C + FM}$
- A more theoretically-based transition by Fuchs is also implemented

PSD: Sectional

- Can't afford to consider all soot sizes
 - 50 nm particle ~ 6 million C's
- Divide the PSD into sections
- Assume uniform size within a section
- Geometrically spaced

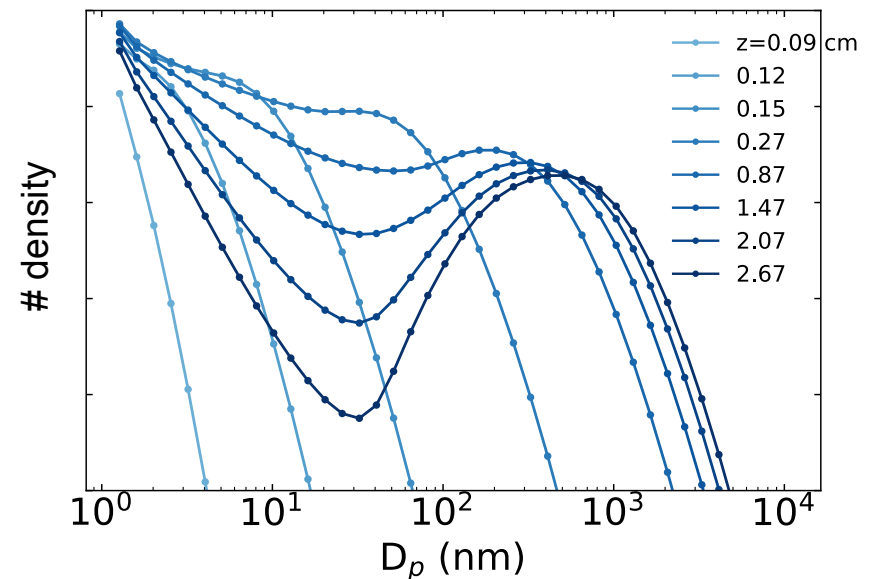
$$m_{min} (F^0, F^1, F^2, \dots, F^{N-1})$$

- Nucleation is into the first bin
- Growth and oxidation are transport in the size coordinate, written in n_i .

$$\frac{dn_{i,g}}{dt} = \frac{k_g A_{i-1} n_{i-1}}{m_i - m_{i-1}} - \frac{k_g A_i n_i}{m_{i+1} - m_i}$$

- Coagulation “lands” between bins
 - Assign to neighbors: conserve # and m

Example: premixed flame evolution



PSD: Method of Moments (MOM)

- Sectional models still require many bins
- Instead, solve for moments of the PSD

$$\frac{dn(m)}{dt} = \dot{S}(n(m))$$



$$M_k = \int m^k n(m) dm$$

$$\frac{dM_k}{dt} = \int m^k \dot{S} dm = \dot{S}_k$$

- The moment source term involves integration over the unknown size distribution $n(m)$.
- This requires a method for closure
- For soot growth, the closure of the fractional moments $M_{k-1/3}$ are needed.

Soot Growth Source

$$\dot{S}_G = -\frac{\partial}{\partial m}(v_g n)$$

$$\dot{S}_{k,G} = -\int_0^\infty m^k \frac{\partial}{\partial m}(v_g n) dm$$

$$\dot{S}_{k,G} = k \int_0^\infty v_g m^{k-1} n dm$$

$$v_g = k_s A$$

$$A = \pi D^2$$

$$m = \rho_s \frac{\pi}{6} D^3$$

$$v_g = k_s \pi \left(\frac{6}{\pi \rho_s} \right)^{2/3} m^{2/3}$$

$$\dot{S}_{k,G} = k_s \pi \left(\frac{6}{\pi \rho_s} \right)^{2/3} k \int_0^\infty m^{k-1/3} n dm$$

$$\dot{S}_{k,G} = k_s \pi \left(\frac{6}{\pi \rho_s} \right)^{2/3} k M_{k-1/3}$$

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Closure Approaches

- **Assumed shape $n(m)$**
 - Monodispersed
 - Lognormal
 - (Power-law + lognormal)
- **Interpolative closure**
 - Interpolate whole order moments to fractional moments
- **Quadrature**
 - QMOM
 - (DQMOM)
 - (CQMOM)

PSD: Assumed Shape $n(m)$

Monodispersed

- Simplest model
- 2 moments
 - M_0 : number density, $\#/m^3$
 - M_1 : soot mass density, kg/m^3
 - $M_1 = \rho Y_s$
 - $f_v = M_1/\rho_s$
- $n(m) = \delta(m - \bar{m})$
$$\bar{m} = \frac{M_1}{M_0} = \frac{\rho Y_s}{n}$$
- Direct integration without closure
- Single (but variable) size
- Special case of QMOM

Lognormal

$$n(m) = \frac{M_0}{\sigma\sqrt{2\pi}} \frac{1}{m} \exp\left(-\frac{\log^2(m/\bar{m})}{2\sigma^2}\right)$$

$$M_k = \int m^k n(m) dm$$

$$M_k = M_0 \bar{m}^k \exp(k^2 \sigma^2 / 2)$$

Write for M_1 , M_2 , solve for $\bar{m}$, σ

$$\sigma^2 = \ln\left(\frac{M_0 M_2}{M_1^2}\right)$$

$$\bar{m} = \frac{M_1^2}{M_0^{3/2} M_2^{1/2}}$$

$$M_k = M_0^{1-3k/2+k^2/2} M_1^{2k-k^2} M_2^{k^2/2-k/2}$$

General fractional moment k

PSD: Assumed Shape $n(m)$

Free molecular collision coefficient has factor

$$\left(\frac{1}{m_1} + \frac{1}{m_2} \right)^{1/2}$$

- This factor precludes direct integration with closure using fraction moments
- Following Pratsinis (1988), assume

$$\left(\frac{1}{m_1} + \frac{1}{m_2} \right)^{1/2} \approx b \left(\frac{1}{m_1^{1/2}} + \frac{1}{m_2^{1/2}} \right)$$

- b is bounded between 0.707 and 1 when m_1 and m_2 approach the same size, and are widely separated, respectively.
- $b=0.854$ gives at worst a 15% error.
- FM coagulation:

$$S_{0,\text{coag}}^{fm} = -Kb (M_1 M_{1/6} + 2M_{2/6} M_{-1/6} + M_{4/6} M_{-3/6})$$

$$S_{1,\text{coag}}^{fm} = 0$$

$$S_{0,\text{coag}}^{fm} = 2Kb (M_1 M_{7/6} + 2M_{8/6} M_{5/6} + M_{10/6} M_{3/6})$$

$$K = c\epsilon_c \left(\frac{\pi k_b T}{2} \right)^{1/2} \left(\frac{6}{\pi} \rho_s \right)^{2/3}$$

Lognormal

$$n(m) = \frac{M_0}{\sigma \sqrt{2\pi}} \frac{1}{m} \exp \left(-\frac{\log^2(m/\bar{m})}{2\sigma^2} \right)$$

$$M_k = \int m^k n(m) dm$$

$$M_k = M_0 \bar{m}^k \exp(k^2 \sigma^2 / 2)$$

Write for M_1 , M_2 , solve for $\bar{m}$, σ

$$\sigma^2 = \ln \left(\frac{M_0 M_2}{M_1^2} \right)$$

$$\bar{m} = \frac{M_1^2}{M_0^{3/2} M_2^{1/2}}$$

$$M_k = M_0^{1-3k/2+k^2/2} M_1^{2k-k^2} M_2^{k^2/2-k/2}$$

General fractional moment k

PSD: Quadrature

General quadrature with function f and weight function W

$$\int_a^b W(x)f(x)dx \approx \sum_{j=1}^N w_j f(x_j)$$

- w_j are weight factors and x_j are abscissas (environments)
- Take $W(x) = n(m)$ for soot.

$$\int_0^\infty n(m)f(m)dm \approx \sum_{j=1}^N w_j f(m_j)$$

- This is equivalent to assuming

$$n(m) = \sum_{j=1}^N w_j \delta(m - m_j)$$

- If w_j and m_j are known, integration is trivial
- Compute w_j , m_j from moments M_k

2 environment Quadrature, 4 moments

$$M_0 = \int m^0 n(m) dm = \sum_{j=1}^2 m_j^0 w_j = w_1 + w_2$$

$$M_1 = \int m^1 n(m) dm = \sum_{j=1}^2 m_j^1 w_j = m_1 w_1 + m_2 w_2$$

$$M_3 = \int m^3 n(m) dm = \sum_{j=1}^2 m_j^2 w_j = m_1^2 w_1 + m_2^2 w_2$$

$$M_4 = \int m^4 n(m) dm = \sum_{j=1}^2 m_j^2 w_j = m_1^3 w_1 + m_2^3 w_2$$

Efficient and stable evaluation using the Wheeler Algorithm (uses LAPACK)

$$\dot{S}_{0,\text{coag}} = \frac{1}{2} \iint \beta(m, \mu) n(m) n(\mu) dm d\mu = -\frac{1}{2} \sum_{j=1}^N \sum_{i=1}^N \beta_{m_j, m_i} w_j w_i$$

PSD: MOMIC

- Frenklach's MOM with interpolative closure.
- Interpolate fractional moments from whole-order moments
 - Interpolate among $\log(M_k)$
 - Positive moments: interpolate among all integer moments
 - Negative moments: interpolate among M_0, M_1, M_2
- Interpolation is done many times for a given set of integer moments.
 - Use a Newton forward polynomial; $O(n)$ operations vs $O(n^2)$
 - (But Barycentric Lagrange Interpolation is better)
- Interpolation among moments works for nucleation, growth, oxidation, and continuum coagulation.
- But special treatment is needed for the FM regime (including treatment of PAH condensation).

Michael Frenklach. Method of moments with interpolative closure. Chemical Engineering Science, 57(12):2229–2239, 2002.

$$S_{0,c}^{FM} = -\frac{K_{FM}}{2} \iint n(m)n(\mu)(m+\mu)^{1/2} (m^{1/6}\mu^{-3/6} + 2m^{-1/6}\mu^{-1/6} + m^{-3/6}\mu^{1/6}) m^0 \mu^0 dm d\mu,$$

$$S_{1,c}^{FM} = 0,$$

$$S_{k \geq 2,c}^{FM} = \frac{K_{FM}}{2} \sum_{j=1}^{k-1} \binom{k}{j} \underbrace{\iint n(m)n(\mu)(m+\mu)^{1/2} (m^{1/6}\mu^{-3/6} + 2m^{-1/6}\mu^{-1/6} + m^{-3/6}\mu^{1/6}) m^k \mu^{r-k} dm d\mu}_{f_{1/2}^{j,k-j}}$$

Grid function $\rightarrow f_{1/2}^{j,k-j}$

Interpolate whole powers to 1/2 power

PSD: MOMIC

$$S_{0,c}^{FM} = -\frac{K_{FM}}{2} \iint n(m)n(\mu)(m+\mu)^{1/2} (m^{1/6}\mu^{-3/6} + 2m^{-1/6}\mu^{-1/6} + m^{-3/6}\mu^{1/6}) m^0 \mu^0 dm d\mu,$$

$$S_{1,c}^{FM} = 0,$$

$$S_{k \geq 2,c}^{FM} = \frac{K_{FM}}{2} \sum_{j=1}^{k-1} \binom{k}{j} \underbrace{\iint n(m)n(\mu)(m+\mu)^{1/2} (m^{1/6}\mu^{-3/6} + 2m^{-1/6}\mu^{-1/6} + m^{-3/6}\mu^{1/6}) m^k \mu^{r-k} dm d\mu}_{f_{1/2}^{j,k-j}}$$

Grid function $\rightarrow$

$$f_0^{x,y} = M_{x-3/6}M_{y+1/6} + 2M_{x-1/6}M_{y-1/6} + M_{x+1/6}M_{y-3/6}$$

$$f_1^{x,y} = M_{x-3/6}M_{y+7/6} + 2M_{x-1/6}M_{y+5/6} + M_{x+1/6}M_{y+3/6} + M_{x+3/6}M_{y+1/6} + 2M_{x+5/6}M_{y-1/6} + M_{x+7/6}M_{y-3/6}$$

$$f_2^{x,y} = M_{x-3/6}M_{y+13/6} + 2M_{x-1/6}M_{y+11/6} + M_{x+1/6}M_{y+9/6} + 2M_{x+3/6}M_{y+7/6} + 4M_{x+5/6}M_{y+5/6} + 2M_{x+7/6}M_{y+3/6} + M_{x+9/6}M_{y+1/6} + 2M_{x+11/6}M_{y-1/6} + M_{x+13/6}M_{y-3/6}$$

$$f_3^{x,y} = M_{x-3/6}M_{y+19/6} + 2M_{x-1/6}M_{y+17/6} + M_{x+1/6}M_{y+15/6} + 3M_{x+3/6}M_{y+13/6} + 6M_{x+5/6}M_{y+11/6} + 3M_{x+7/6}M_{y+9/6} + 3M_{x+9/6}M_{y+7/6} + 6M_{x+11/6}M_{y+5/6} + 3M_{x+13/6}M_{y+3/6} + M_{x+15/6}M_{y+1/6} + 2M_{x+17/6}M_{y-1/6} + M_{x+19/6}M_{y-3/6}$$

- Interpolation using 1, 2, 3, or 4 points
- Many values of x and y
- Exploit symmetry when x=y
- Lots of reused moments
- Create a table: $M_{q/6}$
 - with q = -3, -1, 1, 3, 5 ... 55 handling up to 8 moments
- Continuum requires $M_{p/6}$ with p = -4, -2, 0, 2...
- PAH condensation requires similar treatment, but with different grid function (since m_{Dimer} is used)

SootLib

- SootLib: library of soot models for simulation
- Open source, C++, no external libraries besides LAPACK
- Documented
- Examples: simple interface, premixed flame
- Allows consistent model comparison

Soot Processes

- Nucleation
- Growth
- Oxidation
- Coagulation
- PAH condensation

Particle Distributions

- Monodispersed
- Lognormal
- Quadrature MOM
- MOM-Interpolative Closure
- Sectional

Documentation

SootLib: SootLib

ignite.byu.edu/sootlib_documentation/index.html

Update

 **SootLib** Home Models Namespaces ▾ Classes ▾ Files ▾

SootLib

- Overview
- Dependencies and installation
- Using SootLib
- Examples
- Computational Cost
- Chemistry and Physical Models
- Monodispersed
- Lognormal distribution
- QMOM
- MOMIC
- Sectional
- Bibliography
- Namespaces
- Classes
- Files

SootLib

Overview

Sootlib is an open-source C++ library that computes soot source terms using moment-based particle size distribution models for combustion CFD simulations.

Dependencies and installation

The code is intended to be built and used on Linux-like systems, including MacOS and the Linux subsystem for Windows.

Required software:

- CMake 3.15+
- C++11
- LAPack

Optional software:

- Doxygen (for building documentation)
- graphviz (for Doxygen)
- Catch2 (for building tests; will be locally installed automatically via CMake if `S00TLIB_BUILD_TESTS` is true)

Build and installation instructions

1. Create and navigate into a top-level `build` directory
2. Configure CMake: `cmake ..`
3. Build SootLib: `make`
4. Install SootLib: `make install`

Code Interface

```
#include <iostream>
#include <iomanip>
```

```
#include "sootHeaders.h"
```

```
using namespace std;
using namespace soot;
```

```
int main(int argc, char** argv) {
```

```
    //----- set up and create a soot model
```

```
    nucleationModel *nucl = new soot::nucleationModel_LL();
    growthModel      *grow = new soot::growthModel_LL();
    oxidationModel   *oxid = new soot::oxidationModel_LL();
    coagulationModel *coag = new soot::coagulationModel_FM();
```

```
    size_t nsoot = 4; // number of soot moments (or sections)
```

```
    sootModel_QMOM SM(nsoot, nucl, grow, oxid, coag);
```

```
    //----- set up thermodynamic state variables
```

```
    state S = state(nsoot);
```

```
    double T      = 2100; // temperature in K
    double P      = 101325; // pressure in Pa
    double rhoGas = 0.1; // gas density in kg/m^3
    double muGas  = 1E-5; // gas viscosity in Pa*s
```

```
    vector<double> yGas{3E-4, 0.002, 0.001, 0.05, 0.003, 0.07, 0.1, 0.002};
    vector<double> yPAH{0, 0, 0, 0, 0, 0};
    vector<double> Msoot{0.003, 1.5E-5, 1E-7, 1E-10};
```

```
    S.setState(T, P, rhoGas, muGas, yGas, yPAH, Msoot, nsoot);
```

**Boiler plate;
Collected soot header files
Namespace**

Soot Chemistry, Coagulation Models

PSD model

QMOM, MONO, LOGN, MOMIC, SECT

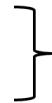
State object

T, P, ρ , μ
y_i: H, H₂, O, O₂, OH, H₂O, CO, C₂H₄
y_{PAH}: C₁₀H₈, C₁₂H₈, C₁₂H₁₀, C₁₄H₁₀,
C₁₆H₁₀, C₁₈H₁₀
Soot variables (M_k or n_{i,section})

Code Interface

```
//----- calculate source terms
```

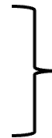
```
SM.setSourceTerms(S);
```



Set source terms, given State S

```
//----- Access soot source terms
```

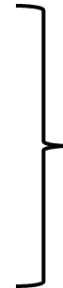
```
double S0 = SM.sources.sootSources[0];  
double S1 = SM.sources.sootSources[1];  
double S2 = SM.sources.sootSources[2];  
double S3 = SM.sources.sootSources[3];
```



Source terms for soot, gas, PAH in sources object

```
//----- Access gas source terms
```

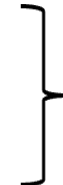
```
double S_O2 = SM.sources.gasSources[(size_t)gasSp::O2];  
double S_O = SM.sources.gasSources[(size_t)gasSp::O];  
double S_H2 = SM.sources.gasSources[(size_t)gasSp::H2];  
double S_H = SM.sources.gasSources[(size_t)gasSp::H];  
double S_OH = SM.sources.gasSources[(size_t)gasSp::OH];  
double S_H2O = SM.sources.gasSources[(size_t)gasSp::H2O];  
double S_CO = SM.sources.gasSources[(size_t)gasSp::CO];  
double S_C2H2 = SM.sources.gasSources[(size_t)gasSp::C2H2];  
double S_C6H6 = SM.sources.gasSources[(size_t)gasSp::C6H6];
```



**Soot-gas source term coupling
Species access by name (enumeration)**

```
//----- Access PAH source terms
```

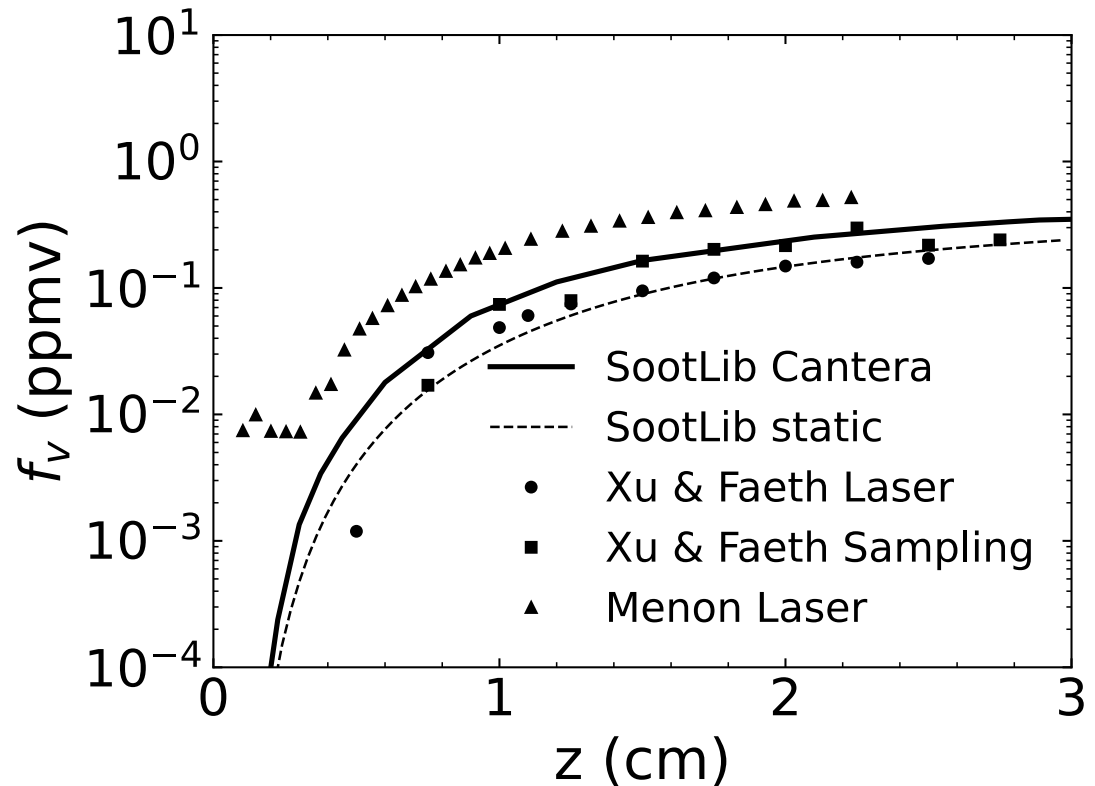
```
C10H8 = SM.sources.pahSources[(size_t)pahSp::C10H8];  
C12H8 = SM.sources.pahSources[(size_t)pahSp::C12H8];  
C12H10 = SM.sources.pahSources[(size_t)pahSp::C12H10];  
C14H10 = SM.sources.pahSources[(size_t)pahSp::C14H10];  
C16H10 = SM.sources.pahSources[(size_t)pahSp::C16H10];  
C18H10 = SM.sources.pahSources[(size_t)pahSp::C18H10];
```



PAH species organized separately

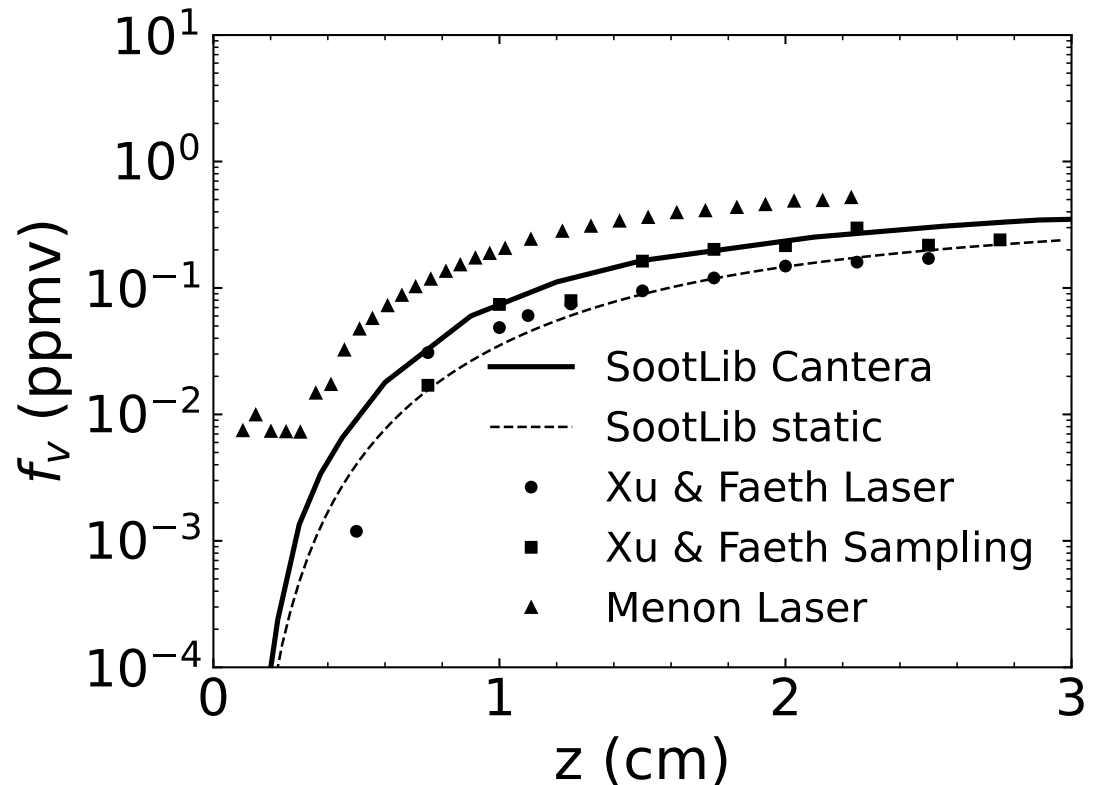
Example: Burner Flame

- Burner stabilized premixed flame
- ISF laminar premixed flame 2
 - C_2H_4 -air
 - $\Phi = 2.34$
 - $v_0 = 6.73 \text{ cm/s}$
- Integrate SootLib into Cantera
 - Modified `flamespeed.cpp`
 - Fixed $T(z)$ profile, experimental
- Models
 - Gas chemistry: gri3.0
 - Nucleation: Lindstedt
 - Growth: Lindstedt
 - Oxidation: Leung
 - Coagulation: FM
 - PSD: Monodispersed

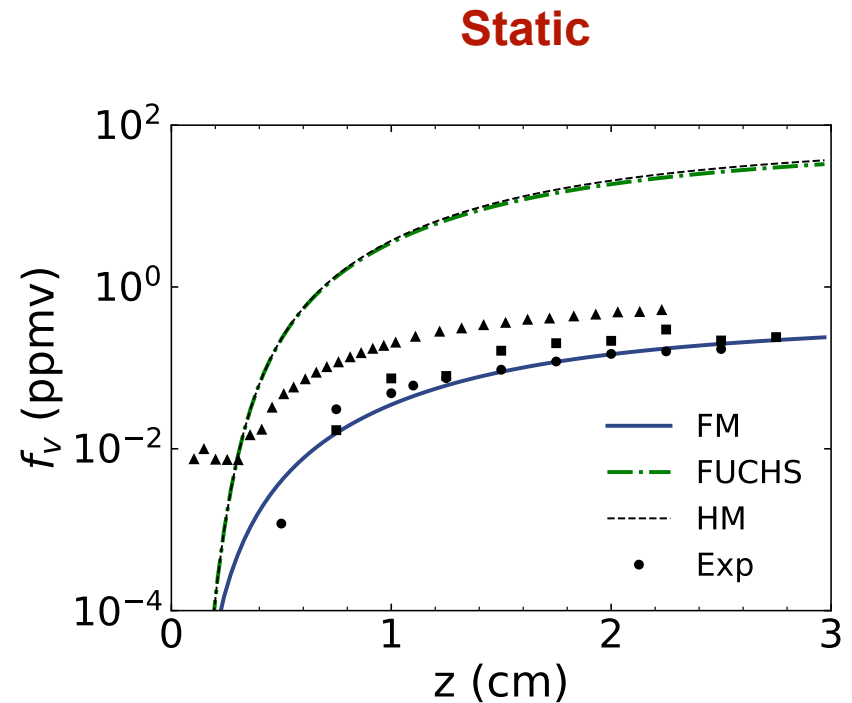
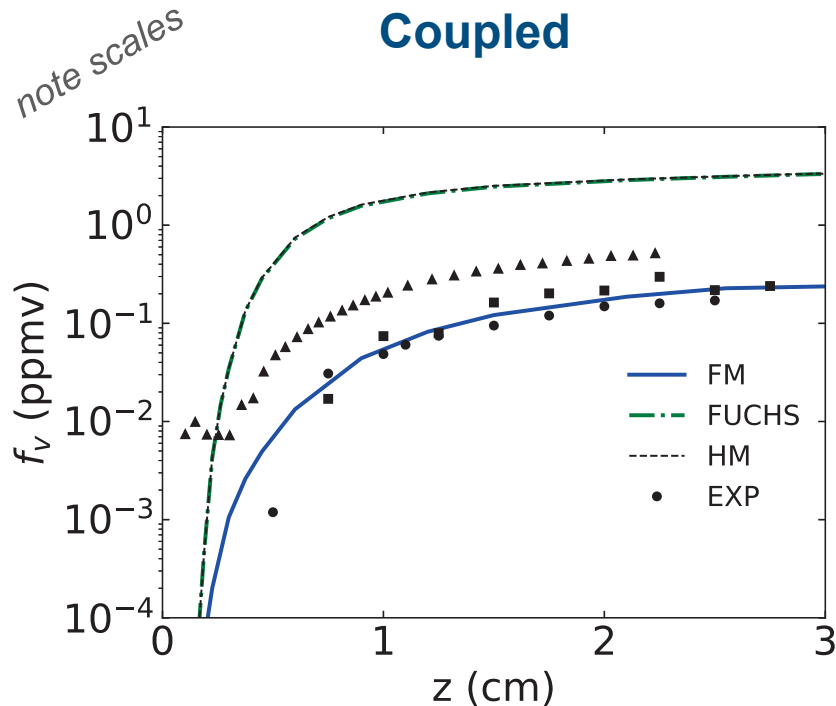


Example: Burner Flame

- Two cases
 - Coupled gas-soot in Cantera
 - Static calculation
- Static calculation
 - Evolves soot using fixed profiles from the (coupled) Cantera run
 - T, ρ, μ, ν, y_i
 - $$\frac{d\hat{M}_k}{dz} = \frac{S_k}{\rho v}$$
$$\hat{M}_k = M_k / \rho$$
 - 2nd order integration
 - Included as a simple code example
- Reasonable agreement with the coupled simulation



Coagulation

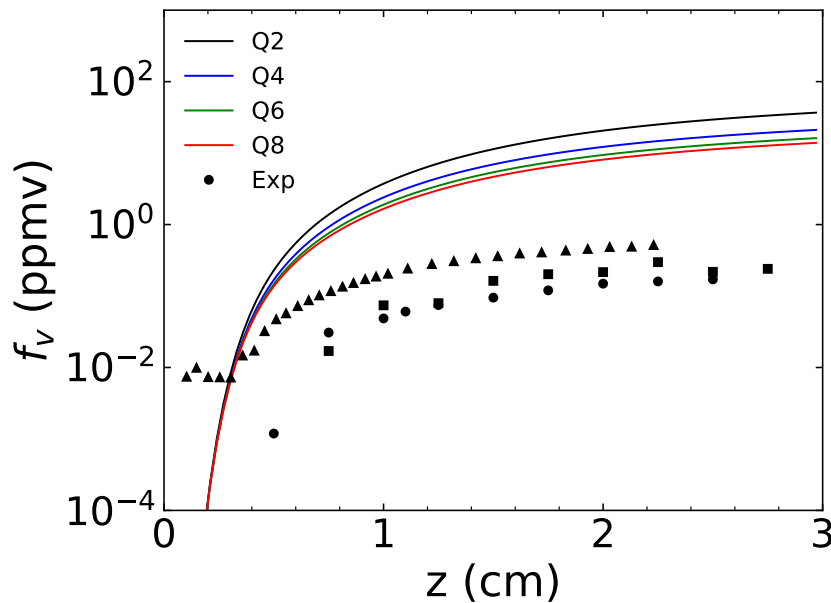


- Large variation with coagulation models used
- Only simple models considered here: Monodispersed, semiempirical chemistry
- HM and Fuchs give nearly identical results
- Lower FM f_v from fewer particles $\rightarrow$ lower surface area for growth
- Soot models should consider effects of all model components

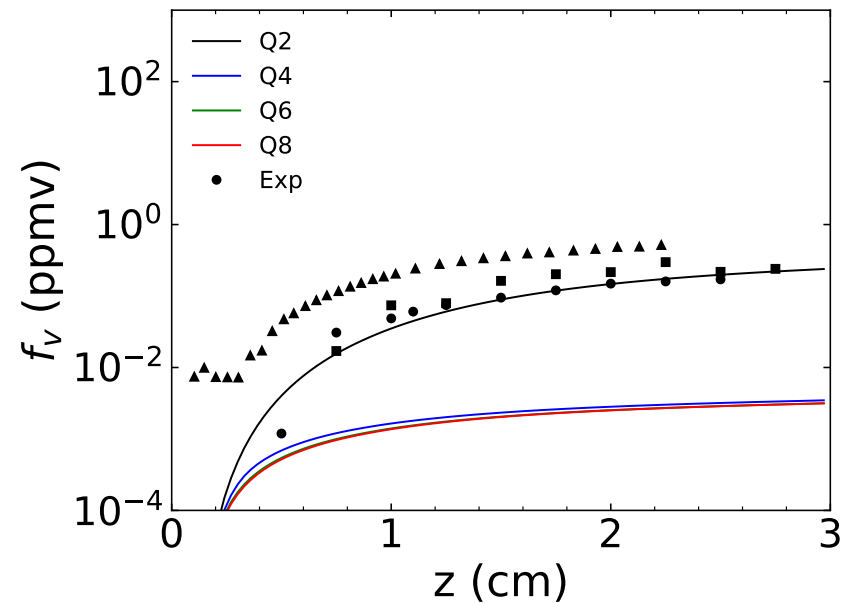
QMOM

Static

HM Coagulation



FM Coagulation

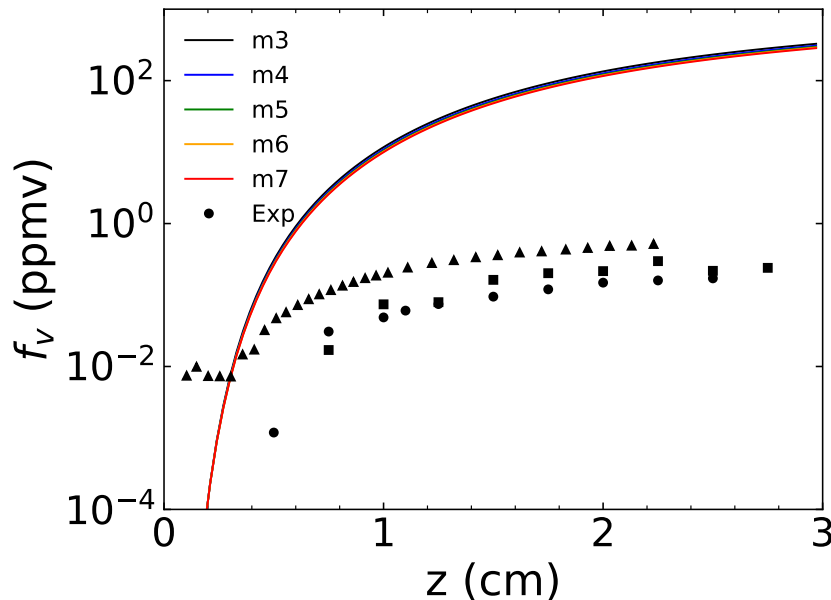


- QMOM shows convergence with increasing environments
 - Q2 has 2 moments, 1 quadrature node (equivalent to monodispersed)
 - Q4 has 4 moments, 2 quadrature nodes
 - Q6 has 6 moments, 3 quadrature nodes
 - Q8 has 8 moments, 4 quadrature nodes
- Q2 $\rightarrow$ Q8 at $z=3$ cm

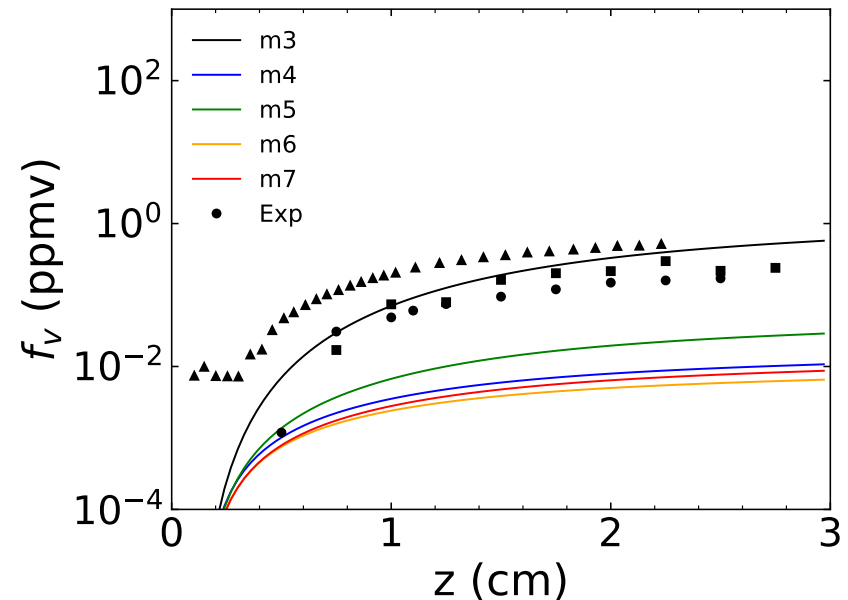
MOMIC

Static

HM Coagulation



FM Coagulation



- FM results show interleaving curves for even and odd number of moments as convergence is approached
- QMOM and MOMIC approach similar converged value for FM.
- MOMIC is higher for HM than QMOM, has less spread.
 - Application of HM is less clean in MOMIC than in QMOM since it is applied on the moment source not “pre-convolution”.
- QMOM is simpler, easier to code, and faster to run.

Github

<https://github.com/BYUignite/sootlib>

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BYUignite licence file 0570110 on Jan 15 414 commits

docs	minor note on LAPack in documentation	2 months ago
examples	updating examples; removing eispac/ext, not working	2 months ago
src	updating CMakeLists.txt for LAPack; tested on mac and linux	2 months ago
tests	BUILD: converting to more robust, project-specific CMake variables; r...	2 years ago
.gitignore	gitignore update	10 months ago
CMakeLists.txt	Documenting source code: formating of top level files done; Doxyfile:...	3 months ago
LICENSE	licence file	2 months ago
README.md	adding documentation link to readme	2 months ago

README.md

SootLib overview

Sootlib is an open-source C++ library that computes soot source terms using moment-based particle size distribution models for combustion CFD simulations. Detailed code documentation is available [here](#)

Dependencies and installation

The code is intended to be built and used on Linux-like systems, including MacOS and the Linux subsystem for Windows.

Required software:

- CMake 3.15+
- C++11

About

Library of soot chemical mechanisms and size distribution representations.

Readme MIT license 0 stars 3 watching 2 forks

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Languages

C++ 94.8% CMake 4.6% C 0.6%

