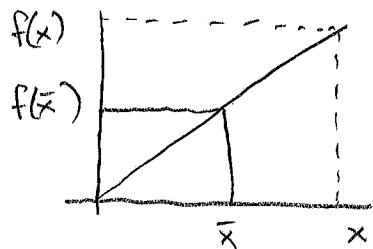


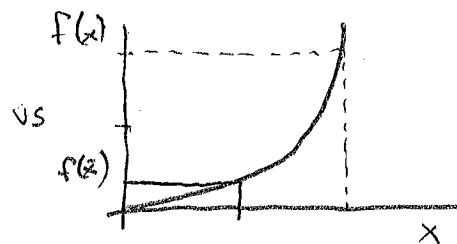
Turbulent Combustion Modelling. - Chemistry.

- Previously Discussed wide Range of Scales in turbulent flows.
- RANS, LES Don't resolve these Scales \rightarrow Models
- Key issues.
 - wide range of scales
 - Nonlinear interactions : $R_{12} \sim e^{-E/RT}$
 - Many species
 - Differential Diffusion
- Turbulence is "random" \rightarrow Deal with Statistics.
 - Mean \bar{V}
 - Mean \bar{T}
 - Mean $\bar{\omega}$, etc.
 - Mean, RMS fluctuations.
- Closure Problem
 - Decompose : $u \rightarrow \bar{u} + u'$
 - Average \rightarrow eqn for \bar{u} , with new terms $\bar{u}'\bar{u}'$
- Species eqn:

$$\frac{\partial \bar{\rho} \tilde{Y}_i}{\partial t} + \nabla \cdot (\bar{\rho} \tilde{v} \tilde{Y}_i) + \nabla \cdot (\bar{\rho} \tilde{v}' \tilde{Y}_i') + \nabla \cdot \bar{f}_i = \bar{R}_i$$
- How to evaluate \bar{R}_i ?
- $A + B \rightarrow C ; R_A = [A][B]k e^{-E/RT}$
- $R_A = R_A(Y_A, Y_B, T)$
- $\bar{R}_A \neq R_A(\bar{Y}_A, \bar{Y}_B, \bar{T})$ Because R is nonlinear

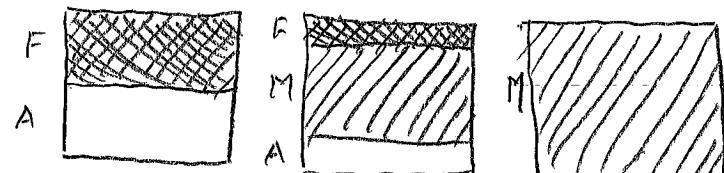


$$\bar{f} = f(\bar{x})$$



$$\bar{f} \neq f(\bar{x})$$

- With fluctuations, we can have a given average, with a wide variety of states
 - Stoic Fuel, Air



$$R=0$$

$$R = \text{intermediate} \quad R = \max$$

Segregated \longrightarrow Mixed

$$* \bar{R} = \iiint R(Y_A, Y_B, T) P(Y_A, Y_B, T) dY_A, dY_B, dT$$

• Need Joint PDF : Multi-D

Unknown

Approaches

- F • Transported PDF
- F • Conditional Moment Closure
- F • Presumed PDF
 - Equilibrium
 - Flamelets
 - ...
- F • EDC
- F • LEM / ODT
- F • Multi-Environment PDF

F \equiv Fluent has a basic implementation.

Presumed PDF Approach - Mixture-fraction based.

(3)

- Observation
 - Typical Combustion Scalars (T, Y_a, ρ) are Strong functions of mixture fraction: ξ

- See Examples

- Result: Eval means as $\bar{\phi}(\xi) = \int_0^1 \phi(\xi) P(\xi) d\xi$
where ϕ is some scalar, like T, Y_a, ρ , radiation absorption coeff., etc.
 - Don't explicitly Transport Species / Reactions.
 - Instead
 - Specify $\phi(\xi)$ (somehow) = Reaction Model
 - Specify $P(\xi)$ (somehow) = Mixing Model

Equilibrium Model.

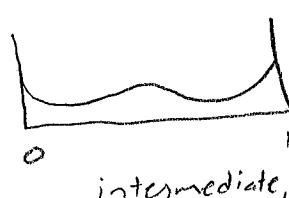
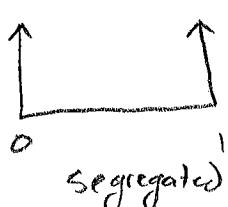
Take $\phi(\xi)$ from equilibrium,

- Discretize ξ into bins (say 100)
- Eval all desired properties as equilibrium in each ξ_i
- Tabulate $\phi_e(\xi_i)$
- Given ξ_e , lookup ϕ_e from Table (interpolate)

$$\bar{T} = \int_0^1 T(\xi) P(\xi) d\xi$$

Need $P(\xi)$

- For given ξ , limits of $P(\xi)$:



- PDF is specified by (1) Form

$$\left. \begin{array}{c} (2) \bar{\xi} \\ (3) \sigma_{\xi}^2 \end{array} \right\} \rightarrow \bar{\xi}', \sigma_{\xi'}^2$$

Clipped Gaussian



- Clip tails $\pm \infty$, add δ functions at 0, 1

$$P(\xi) = \begin{cases} \alpha_0 = \frac{1}{2}(1 - \operatorname{erf}(\frac{\bar{\xi}}{\sqrt{2}\sigma_{\xi}^2})) & : \bar{\xi} = 0 \\ \frac{1}{\sqrt{2\pi}\sigma_{\xi'}^2} \cdot \exp\left[-\frac{(\xi - \bar{\xi})^2}{2\sigma_{\xi'}^2}\right] & : 0 < \bar{\xi} < 1 \\ \alpha_1 = \frac{1}{2}(1 - \operatorname{erf}((\bar{\xi} - 1)/\sqrt{2\sigma_{\xi'}^2})) & : \bar{\xi} = 1 \end{cases}$$

β -PDF

- Similar, but more natural

$$P(\xi) = \xi^{a-1} (1-\xi)^{b-1} \cdot \frac{\Gamma(a+b)}{\Gamma(a)\Gamma(b)}$$

$$a = \bar{\xi} \left[\frac{\bar{\xi}(1-\bar{\xi})}{\sigma_{\xi}^2} - 1 \right] = \bar{\xi} (\sigma_{\xi, \text{max}}^2 / \sigma_{\xi}^2 - 1)$$

$$b = a(1-\bar{\xi})/\bar{\xi}$$

Γ is the gamma function.

See Examples.

For Equilibrium Model

- Build a 2-D Table $\bar{\phi} = \bar{\phi}(\bar{\xi}, \sigma_{\xi}^2)$

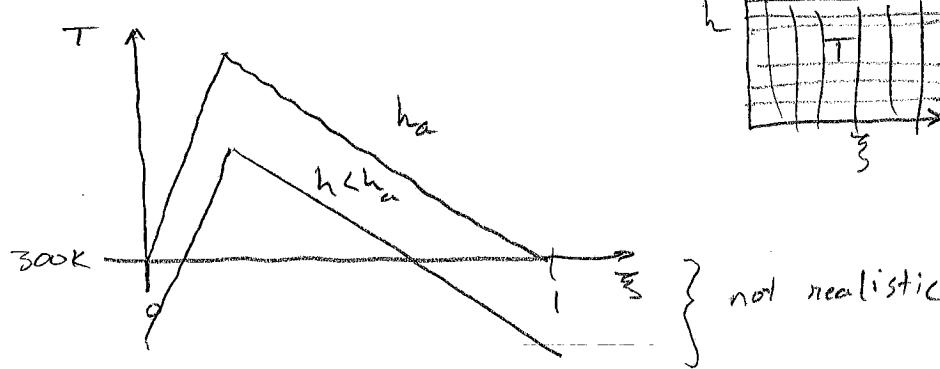
- computed as $\bar{\phi} = \int \phi(\xi) P(\bar{\xi}, \sigma_{\xi}^2) d\xi$

- CFD Solves Transport equations for $\bar{\xi}, \sigma_{\xi}^2$
→ Lookup $\bar{\phi}$ as needed

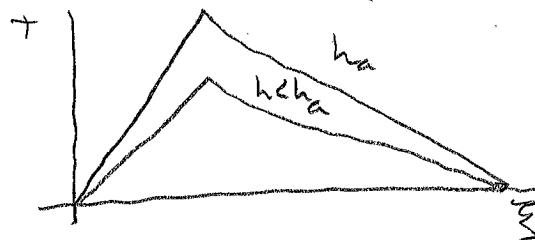
* Note, the Table is needed for closing terms in the resulting Transport eqns (like $\bar{\rho}$), not e.g. \bar{Y}_i

Head loss.

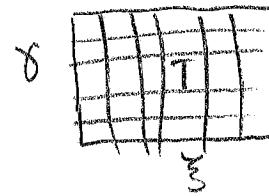
- Via Radiation, Wall transfer
- Code Solves Enthalpy, but it is inconvenient to tabulate chemistry in terms of h :



- instead, use $\gamma = \frac{h_a(\xi) - h}{h_s(\xi)}$



• Pure fuel, air
have $h_s \rightarrow 0$
 $\Rightarrow h \rightarrow h_a$



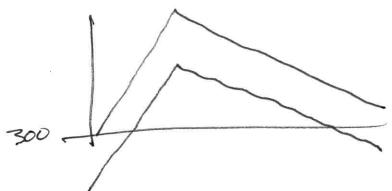
. $\bar{\phi} = \bar{\phi}(\bar{\xi}, \sigma_{\xi}^2, \gamma)$ Table.

. Transport $\bar{\xi}, \sigma_{\xi}^2, \bar{h}$

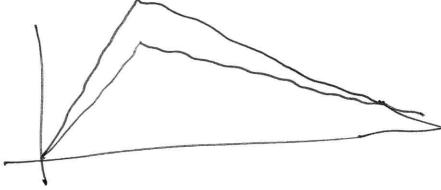
. Lookup $\bar{\phi}(\bar{\xi}, \sigma_{\xi}^2, \gamma)$ as needed

$$\begin{aligned}
 * \text{ Compute } \bar{\phi} \text{ as } \bar{\phi} &= \iint \phi(\xi, \gamma') P(\xi, \gamma') d\xi d\gamma' \\
 &\approx \iint \phi(\xi, \gamma) P(\xi) P(\gamma) d\xi d\gamma' \\
 &\approx \iint \phi(\xi, \gamma) P(\xi) \delta(\gamma' - \gamma) d\xi d\gamma' \\
 &= \int \phi(\xi, \gamma) P(\xi) d\xi
 \end{aligned}$$

Heat loss Note

- When using an equilibrium model, the various ξ are independent, unlike a 1-D laminar flamelet.
- Then, to integrate over the β -PDF, we need some appropriate measure of h that applies across the ξ space.
- Cannot use a uniform value of h (or \underline{h})


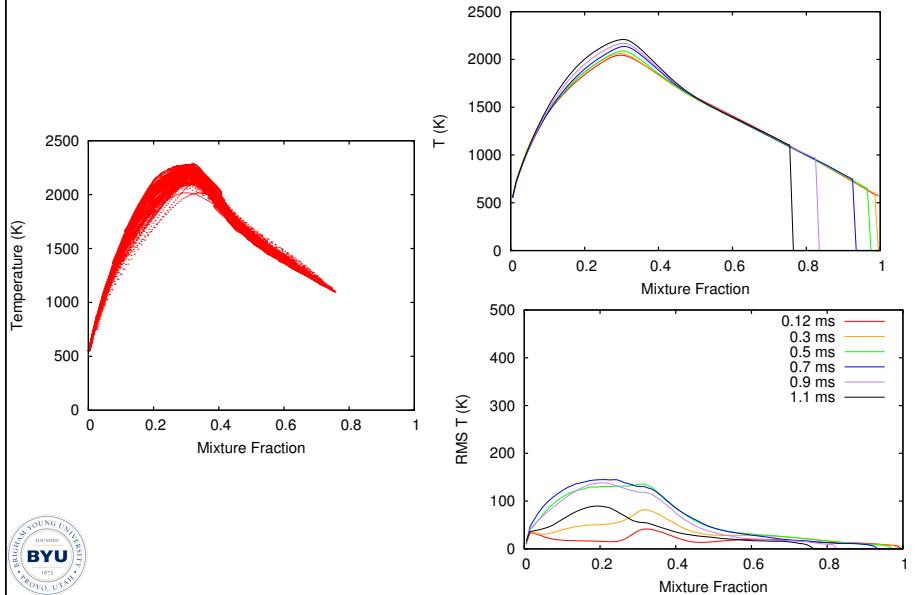
since we would go "negative" on the boundaries.
e.g. uniform \underline{h}
- γ Scales \underline{h} by h_s which vanishes on the edges.

$$h = h_0 - h_s \cdot \gamma \quad \rightsquigarrow \text{as } h_s \rightarrow 0 \quad h \rightarrow h_0$$


e.g. uniform γ .
- Other possibilities could work:
 - Radiate to "cold" surroundings for some Δt .

 $T_{\text{mix}}(\xi)$
 Do this for all ξ for same Δt .

Jet Temperature Profiles



Experimental Profiles

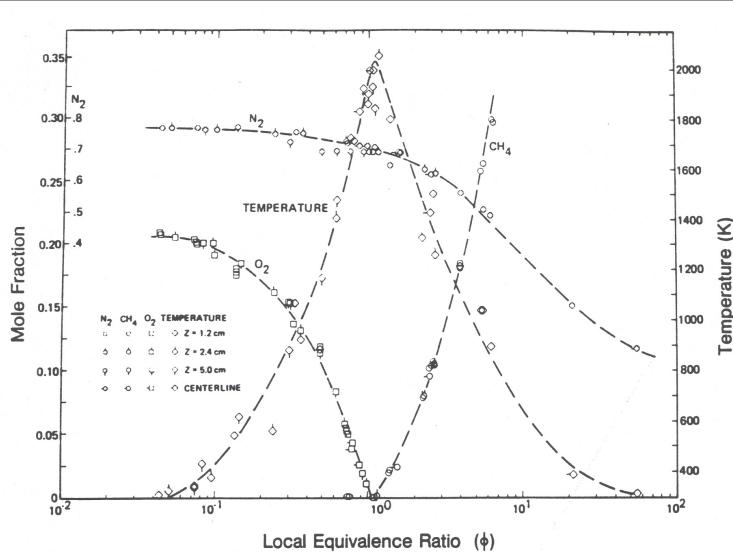
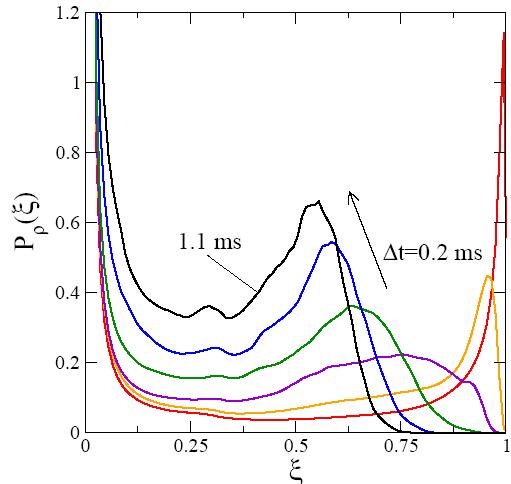


Figure 11.4. Measured temperature and reactant concentration as a function of the local equivalence ratio in a laminar methane diffusion flame. (Figure used with permission from Mitchell *et al.*, 1980.)



Mixture Fraction PDFs

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Sample Equilibrium Table

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